

## CHEMICAL SYNTHESIS OF D-ERYTHRO-SPHINGOSINE-1-PHOSPHATE, AND ITS INHIBITORY EFFECT ON CELL MOTILITY

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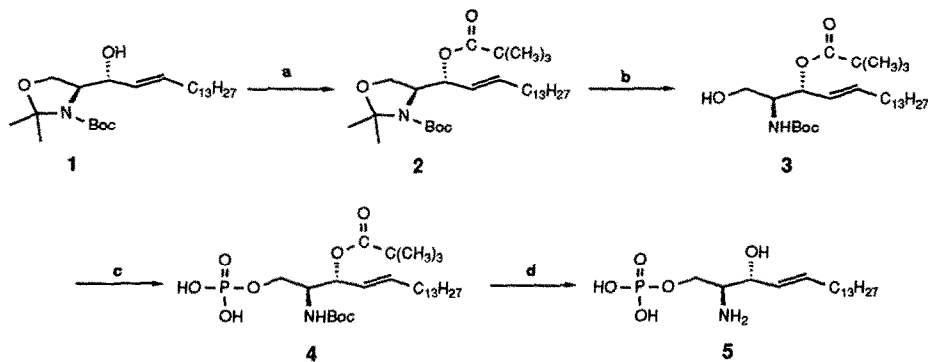
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**Abstract.** The first chemical synthesis of D-erythro-sphingosine-1-phosphate (which occurs naturally) is described. This synthetic product had an inhibitory effect on motility of mouse melanoma B16/F1 cells in an *in vitro* assay system.

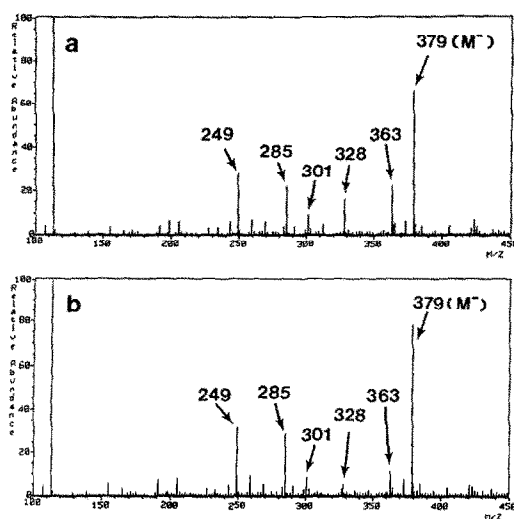
Sphingosine-1-phosphate (SPN-1-P) has been known for many years as an intermediate product during degradation of sphingosine (SPN) by SPN kinase to ethanolamine-1-phosphate and a long-chain aldehyde (e.g., palmital) by a pyridoxal phosphate-dependent lyase reaction<sup>1</sup>. The biological significance of SPN-1-P has been reported recently<sup>2</sup>. However, its physiological function in cells remains unclear, except for its Ca<sup>2+</sup> mobilizing activity in some cells<sup>2c</sup>. Recently, we demonstrated that SPN-1-P inhibits motility of melanoma cells at a very low concentration (10 nM), at which SPN, N,N-dimethyl-SPN, and N,N,N-trimethyl-SPN have no inhibitory effect<sup>3</sup>. Furthermore, SPN-1-P is far less cytotoxic than these other three compounds, and does not inhibit protein kinase C. These biological findings suggest that SPN-1-P may act as an agent for prevention of tumor cell metastasis and inflammatory processes, both of which are highly dependent on cell motility. Unlike SPN and its N-methylated derivatives, the chemical synthesis of SPN-1-P has not been reported. The only known method for preparation of SPN-1-P is by treatment of sphingosylphosphorylcholine (SPC) with phospholipase D from *Streptomyces chromofuscus*<sup>4</sup>, which gives a mixture of D-erythro and L-threo isomers. During design and synthesis of cell motility inhibitors derived from SPN-1-P, we developed a chemical synthesis of D-erythro-SPN-1-P, the naturally-occurring isomer.

The chemical synthesis of SPN-1-P is quite straightforward. In order to selectively phosphorylate the primary hydroxyl group of SPN, its amino and allylic hydroxyl groups must be protected. We chose the protected D-erythro-olefinic alcohol 1

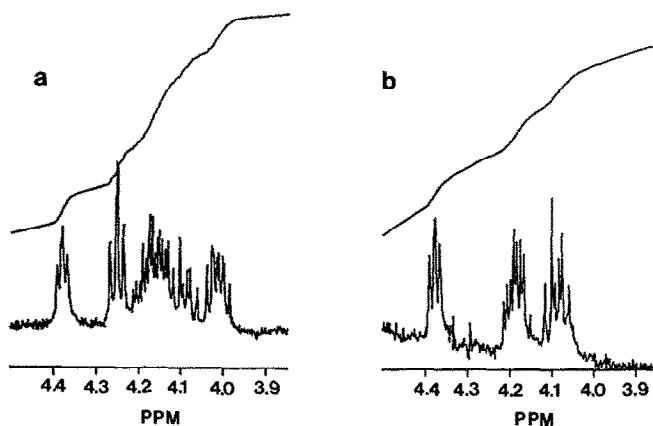
## Scheme I



(a) pivaloyl chloride, pyridine; (b) p-TsOH, CH<sub>3</sub>OH; (c) (i) POCl<sub>3</sub>, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>, (ii) H<sub>2</sub>O, pyridine, dioxane; (d) (i) nBu<sub>4</sub>N<sup>+</sup>OH<sup>-</sup> (aq), dioxane; Amberlite IR-120 (H<sup>+</sup>), (ii) 50% TFA/CH<sub>2</sub>Cl<sub>2</sub>.



**Figure 1.** Negative-ion FAB mass spectra (DMIX as a matrix) of sphingosine-1-phosphate from (a) sphingosylphosphocholine with phospholipase D; (b) chemical synthesis.



**Figure 2.** Portions of  $^1\text{H}$ -NMR spectra (500 MHz) of sphingosine-1-phosphate from (a) sphingosylphosphocholine with phospholipase D,  $\delta$  (ppm) 3.97-4.22 (4H, m,  $\text{POCH}_2$ ), 4.25, 4.38 (2H, t,  $\text{CH}(\text{OH})$ ); (b) chemical synthesis,  $\delta$  (ppm) 4.09, 4.18 (2H, m,  $\text{POCH}_2$ ), 4.38 (1H, t,  $\text{CH}(\text{OH})$ ), taken in methyl- $^{12}\text{C}$ - $\text{d}_3$ -alcohol- $\text{d}$ -acetic- $\text{d}_3$ -acid- $\text{d}$  8:2 (v/v).

( $[\alpha]_{\text{D}}^{25} -25.9^\circ$  ( $c = 1.43$ ,  $\text{CHCl}_3$ ); lit.<sup>5b</sup>  $-25.2^\circ$  ( $c = 0.215$ ,  $\text{CHCl}_3$ )), readily available from L-serine by the known method<sup>5</sup>, as the starting material. We also chose the acid-stable pivaloyl group for protection of the allylic-OH group, which could be left intact through modification of the 1-OH group, with no C-3 to C-1 migration<sup>6</sup>. Synthesis of SPN-1-P is summarized in Scheme 1. Esterification of **1** (6 eq.  $(\text{CH}_3)_3\text{CCOCl}$  in dry pyridine,  $0 \rightarrow 25^\circ\text{C}$ , 5 h) gave **2** (98%) as a colorless oil ( $[\alpha]_{\text{D}}^{25} -25.7^\circ$  ( $c = 1.22$ ,  $\text{CHCl}_3$ )). Selective deprotection of the 1-OH group (1.3 eq.  $\text{TsOH} \cdot \text{H}_2\text{O}$  in methanol, 5 h) produced **3** (70%) as a colorless oil ( $[\alpha]_{\text{D}}^{25} -12.9^\circ$  ( $c = 1.20$ ,  $\text{CHCl}_3$ )). Phosphorylation of the primary hydroxyl group of **3** (4 eq.  $\text{POCl}_3$ , 4.5 eq.  $\text{Et}_3\text{N}$  in dry  $\text{CH}_2\text{Cl}_2$ ,  $0 \rightarrow 25^\circ\text{C}$ , 2 h), followed by hydrolysis ( $\text{H}_2\text{O}$ , pyridine in dioxane, 1.5 h), gave the precursor of SPN-1-P, **4** (53%). Finally, sequential deprotection of **4** (excess amount 40% wt.  $\text{nBu}_4\text{N}^+\text{OH}^-$  (aq) in dioxane, 4 h, then Amberlite IR-120( $\text{H}^+$ ); 50% TFA/ $\text{CH}_2\text{Cl}_2$ , 0.5 h) yielded the naturally-occurring D-erythro-SPN-1-P, **5**, as a white solid in 78% yield after chromato-

graphy on silica gel (nBuOH/H<sub>2</sub>O/AcOH, 5:1:1). Mobility of this synthetic product on silica-gel TLC (nBuOH/H<sub>2</sub>O/AcOH, 6:1:1) and HPTLC (CHCl<sub>3</sub>/MeOH/H<sub>2</sub>O, 65:35:8) was indistinguishable from that of enzymatically-derived material prepared by the method of van Veldhoven *et al.*<sup>8</sup> Negative ion FAB-MS (Figure 1) revealed that the

**TABLE I.** Inhibitory effects of chemically and enzymatically synthesized SPN-1-P on F1 melanoma cell motility.

Compound	Concentration ( $\mu$ M)	% migration of F1 cells <sup>a</sup>
Control		100 $\pm$ 32
SPN-1-P <sup>b</sup>	5	34 $\pm$ 9
	1	8 $\pm$ 2
	0.1	6 $\pm$ 2
	0.01	12 $\pm$ 7
	0.001	82 $\pm$ 44
SPN-1-P <sup>c</sup>	5	41 $\pm$ 10
	1	10 $\pm$ 1
	0.1	12 $\pm$ 1
	0.01	17 $\pm$ 2
	0.001	122 $\pm$ 39
SPN	5	12 $\pm$ 3
	1	78 $\pm$ 10
	0.1	90 $\pm$ 22

<sup>a</sup> mean $\pm$ S.D. (n= 3 or 4). Control value (defined as 100%) represented 2.2 $\times$ 10<sup>4</sup> migrated cells. See footnote 9 for experimental details.

<sup>b</sup> Enzymatically synthesized.

<sup>c</sup> Chemically synthesized.

synthetic product gave the correct molecular ion, and its fragmentation pattern is identical to that of enzymatically-derived material. The  $^1\text{H}$ -NMR (500 MHz) spectrum (Figure 2) showed only one set of resonances in H-1 and H-3, indicating that there was no racemization.

Synthetic SPN-1-P was tested *in vitro* for inhibition of chemotactic motility of mouse melanoma B16/F1 cells, using a Matri-gel-coated transwell assay system<sup>9</sup>. The results (summarized in Table I) show that both chemically- and enzymatically-prepared SPN-1-P have a similar dose-dependent inhibitory effect on cell motility. SPN-1-P at a concentration of 10 nM blocked penetration of F1 cells through the Matri-gel-coated filter, while a much higher concentration (5  $\mu\text{M}$ ) was required for SPN.

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- 7 All intermediates were purified by silica-gel chromatography. Their structures were supported (unless otherwise indicated) by  $^1\text{H}$ -NMR and low resolution FAB-MS.
- 8 In each system, SPN-1-P was positive with orcinol, ninhydrin, and phosphate sprays.
- 9 Chemical motility of F1 cells in a Matri-gel-coated transwell chamber was assayed as follows. The upper chamber filter was precoated with 1  $\mu\text{g}$  Matri-gel. F1 cells suspended in DMEM with 2% fetal calf serum ( $5 \times 10^4$  cells/ml) were placed on the filter and incubated for 30 min in a  $\text{CO}_2$  incubator. The upper chamber with attached cells was fitted onto the lower chamber containing 0.6 ml of splenic stromal cell-derived conditioned medium as a chemoattractant, and incubated for 20 h to allow the cells to penetrate the filter. After incubation, cells remaining in the upper chamber were wiped off with a cotton swab, and cells migrating to the lower side of the filter were fixed in methanol and stained with 0.05% toluidine blue. The stain was solubilized in 10% acetic acid, and  $\text{OD}_{630}$  was measured with an ELISA reader. A linear relationship was observed between number of cells migrating to the lower chamber and  $\text{OD}_{630}$ .